Tetrahedron 66 (2010) 9840-9848

Contents lists available at ScienceDirect

Tetrahedron

N -Carbamate α -aminoalkyl-p-tolylsulfones—convenient substrates in the nitro-Mannich synthesis of secondary N-carbamate protected syn-2-amino-1-nitroalkanephosphonates

Roman B1aszczyk, Anna Gajda, Stefan Zawadzki, Ewelina Czubacka, Tadeusz Gajda *

Institute of Organic Chemistry, Faculty of Chemistry, Technical University of Lodz, Zeromskiego St. 116, 90-924 Lodz, Poland

article info

Article history: Received 7 July 2010 Received in revised form 5 October 2010 Accepted 25 October 2010 Available online 29 October 2010

Keywords: Aza-Henry reaction Diethyl nitromethanephosphonate a-Amido sulfones PTC reaction Aminonitrophosphonates 1,2-Diaminoalkanephosphonates

ABSTRACT

An efficient one-pot synthesis of secondary N-carbamate protected syn- β -amino- α -nitroalkanephosphonates using diethyl nitromethanephosphonate and N-Boc or N-Cbz imines, generated in situ from stable N-Boc or N-Cbz a-aminoalkyl-p-tolylsulfones has been developed under PTC conditions. A model enantioselective version of this reaction is also described. Enantioselectivity up to 67% ee is achieved using a chiral thiourea catalyst derived from a cinchona alkaloid. Completely stereoselective conversion of the title compounds into partially N-carbamate protected syn-1,2-diaminoalkanephosphonates has also been elaborated.

2010 Elsevier Ltd. All rights reserved.

1. Introduction

The aza-Henry reaction, $1\frac{1}{7}$ $1\frac{1}{7}$ $1\frac{1}{7}$ known also as the nitro-Mannich reaction, is usually base promoted addition of nitro compounds toNprotected imines, often being employed in the asymmetric synthesis of vicinal diamines and α , β -diamino acids.^{[1b,e,g,2,3a](#page-7-0)-[c,h](#page-7-0)-[j,l,4b,c,5d,6,8,9](#page-7-0)} There are however, only isolated examples of the nitro-Mannich approach to the preparation of β -amino- α -nitro- and α , β -diaminophosphonates 10^{-12} 10^{-12} 10^{-12} 10^{-12} utilizing 1-nitroalkanephosphonates and N-protected imines as starting materials. In the late 1970s Petrov et al.[12](#page-8-0) described the addition of nitromethane- and 1-nitroethanephosphonates to N-benzylidenebenzeneamine leading to N-phenyl analogues of aminonitrophosphonates. Recently, during our investigations, Johnston et al.^{[10](#page-8-0)} published a highly diastereoand enantioselective addition of dialkyl 1-nitroethanephosphonates to N-Boc imines, catalyzed by chiral Brønsted acids, which allows obtaining anti-2-amino-1-nitroethanephosphonates with a quaternary center next to a phosphorus atom. Until now however, the reaction of diethyl nitromethanephosphonate with N-carbamate α aminoalkyl-p-tolylsulfones, as convenient precursors of unstable Ncarbamate imines, leading to the title compounds has not been described. This prompted us to present here our contribution as a new, practical synthesis of secondary 1-amino-2-nitroalkaneand 1,2-diaminoalkanephosphonates utilizing diethyl n-itromethanephosphonate^{[13](#page-8-0)} (1) and N-carbamate α -aminoalkyl-ptolylsulfones 2 as starting materials.

2. Results and discussion

It is well documented that N-Boc and N-Cbz protected α -ami-noalkyl-p-tolylsulfones^{[1d,14](#page-7-0)–[16](#page-7-0)} 2 can be considered as stable, crystalline, and easy to handle equivalents of N-Boc and N-Cbz imines. Not surprisingly, nucleophilic additions of deprotonated nitroalkanes to N -carbamate imines generated in situ from the α -amido sulfones, mentioned above by base-induced elimination, have recently been the subject of extensive studies. $4i,5a,b,d,7$ In this context, easy to perform and low demanding as far as reactions conditions are concerned, phase transfer catalysis $(PTC)^{1f,17}$ $(PTC)^{1f,17}$ $(PTC)^{1f,17}$ approaches to the aza-Henry reaction seemed to be especially beneficial. After preliminary experimentations, standard liquid-solid PTC conditions using potassium carbonate as a base for simultaneous deprotonation and elimination, tetrabutylammonium bromide (TBAB) as catalyst and toluene as solvent were selected for the reaction of diethyl nitromethanephosphonate (1) with N-carbamate protected α -aminoalkyl-p-tolylsulfones 2 ([Scheme 1](#page-1-0)).

^{*} Corresponding author. Tel.: $+48$ 42 6313146; fax: $+48$ 42 6365530; e-mail address: tmgajda@p.lodz.pl (T. Gajda).

^{0040-4020/\$ -} see front matter \odot 2010 Elsevier Ltd. All rights reserved. doi:10.1016/j.tet.2010.10.072

Scheme 1. Reagents and conditions: (i) K_2CO_3 solid (5 equiv), TBAB (10 mol %), toluene, rt, 26 h; (ii) aq KHSO₄

As shown in Scheme 1, a vigorously stirred mixture of 1, N-Boc protected α -aminoalkyl-p-tolylsulfones **2a–f** (1.03 equiv), potassium carbonate (5 equiv) and TBAB (10 mol %) in toluene afforded, after 26 h at room temperature, the mixture of syn- and anti-adducts $3a-f$ in good yields (79–88%) after flash chromatography. The results are summarized in Table 1.

The results given in Table 1 indicate that adducts 3 are formed with low and comparable syn-diastereoselectivity (syn-3a-f/anti-3a $-f=56/44-64/36$). The reaction also works for N-Cbz protected α -aminoalkyl-p-tolylsulfones 2g-h as imine surrogates, for which,

Table 1

Diethyl β -amino- α -nitrophosphonates **3a-h** prepared

Entry 3		\mathbb{R}				PG Yield ^{a,b} (%) syn/anti ^c syn-3 Yield ^d (%) ³¹ P NMR, δ		
							syn	anti
		3a Ph	Boc 87		59/41	40 ^e		10.36 11.02
$\overline{2}$		3b p -MeOC ₆ H ₄		83	63/37	39		10.08 10.79
3		3c p -ClC ₆ H ₄		88	64/36	41		10.10 10.76
4		3d 2-Furyl		87	63/37	41		9.88 10.55
5		3e Me		88	61/39	$\qquad \qquad -$		11.09 11.27
6	$3f$ Et			79	56/44	$\overline{}$		11.00 11.29
7		3g Ph	Cbz 71		53/47	28 ^f		10.17 10.93
8		3h Me		84	57/43			10.67 10.81

^a All reactions were carried out on a 2 mmol scale.

Overall yield of pure syn- and anti-adducts 3a-h isolated after flash chromatography.

^c Diastereomeric ratio measured by 31 P NMR of the crude products.

Yields of pure syn-3, after crystallization from Cl_4 .

e Second crop of crystals was isolated in 19% yield and consisted of *anti*-3a mainly
(*syn/anti*=25/75).

¹ The mixture of syn-and *anti-3a* (syn/anti-10/00) was isolated in 27% with firm

The mixture of syn- and anti-3g (syn/anti=10/90) was isolated in 27% yield from the mother liquor.

the mixtures of N-Cbz protected syn- and anti-adducts 3g and 3h are isolated in much the same yields and with similar syn-diastereoselectivity (Table 1, entries $7-8$). Facile epimerization on the carbon atom C-1, next to the nitro group, in adducts 3 and/or retro addition/re-addition occurring under the reaction conditions, seem to be responsible for the low diastereoselectivity of the above mentioned reactions.[†] We found, additionally, that stirring the mixture of 1 and 2a or 1, and 2g for 10 days under the same PTC conditions, resulted, within the experimental error, in the same diastereomeric ratios of the final adducts 3a or 3g, respectively.

The evaluation of the scope of the reaction shows that the method is applicable to aromatic, heteroaromatic and aliphatic N-Boc, and N-Cbz protected α -aminoalkyl-p-tolylsulfones 2. Pure syn-3a-d and syn-3g could easily be isolated in $28-41\%$ yields from the mixture of $3a-d$ and $3g$ via simple crystallization from carbon tetrachloride (Table 1, entries $1-5$ and 7, respectively). Pure aliphatic derived syn-adducts $3e$ -f and $3h$, which are formed as viscous oils, could not be separated from the mixture either by crystallization or chromatography.

The mother liquors of adducts 3, obtained after crystallizations, consisted mainly of anti-diastereomers 3, and these mixtures slowly enriched themselves in syn-3 after prolonged standing at room temperature.

Attempted widening of the scope of the above nitro-Mannich reaction on diethyl 1-nitroethanephosphonate,^{[18](#page-8-0)} a homologue of 1, was unsuccessful. Starting 1-nitroethanephosphonate was always recovered from the reaction mixture.

Determination of the relative syn/anti configuration of adducts 3 based on NMR analysis of diagnostic vicinal coupling constants $({}^3\!J_{\rm HH}, \, {}^3\!J_{\rm CP})$ failed. In almost all of the adducts diagnostic protons (H_1 and H_2) appeared as complex multiplets or the differences in the values of their vicinal coupling constants $^3J_{\rm HH}$ were too small to be diagnostic. Therefore, the relative $syn/anti$ configurations of 3 were established by conversion of selected compounds 3 into the appropriate imidazolidine-2-thiones 5 (vide infra). However, the phosphorus chemical shifts of 3 were consistent with the given diastereomer. In the ³¹P NMR spectra of all of nitrophosphonates 3 the signals of the syn-isomers appeared $0.14-0.74$ ppm upfield relative to those of the anti-isomers (Table 1).

Having established the synthesis of N-carbamate protected 2-amino-1-nitroalkanephosphonates 3, we focused our attention on their conversion into partially protected 1,2-diaminoalkanephosphonates 4. Several 1,2-diaminoalkanephosphonic acids act as leucine aminopeptidase inhibitors.^{[19](#page-8-0)} However, the number of known α , β -diaminoalkanephosphonates is limited and only a few routes to both racemic or enantioenriched compounds have been reported till now.[10,20,21](#page-8-0)

Sodium borohydride/nickel(II) chloride^{[22](#page-8-0)} system in methanol was used as the reductant of choice for the nitro group. Such reductions occurred under very mild conditions, and they were shown to proceed with retention of configuration at the stereogenic center on the carbon atom next to the nitro group.^{31,4c,h,6b,d,e,23}

As shown in Scheme 2, solid NaBH₄ was added at -30 °C to the solution of nitrophosphonates 3 and nickel(II) chloride hexahydrate in methanol. The mixture was allowed to warm to 0 \degree C, and desired 1,2-diaminophosphonates 4 were isolated in moderate yields $(43–75%)$, after quenching with aq ammonium chloride followed by flash chromatography. The approach mentioned above enables a twostep synthesis of secondary 1,2-diaminophosphonate esters [\(Table 2\)](#page-2-0).

As shown in [Table 2](#page-2-0), syn-1,2-diaminophosphonates $4a-d$ and 4g were only obtained from the reduction of pure syn-adducts

$$
(EtO)2P\begin{array}{c}\nP\downarrow R \\
R\n\end{array}\n\begin{array}{ccc}\n1 & 0 & NHPG \\
13-75\% & (EtO)2P\begin{array}{c}\nR \\
R\n\end{array}\n\end{array}
$$
\n3a-h\n3,4 PG = Boc; R = Ph (a), p-MeOC₆H₄ (b), p-CC₆H₅ (c), 2-furyl (d), Me (e), Et (f)\n3,4 PG = Cbz; R = Ph (g), Me (h)

Scheme 2. Reagents and conditions: (i) NaBH₄ (10 equiv), NiCl₂×6H₂O (1.05 equiv), MeOH, -30 °C, 5 min, 0 °C; (ii) aq NH₄Cl, 0 °C.

3a-d and 3g [\(Table 2,](#page-2-0) entries 1, $3-5$, and 8, respectively). In turn, the mixture of syn- and anti-adducts $3a$ and $3e-h$ afforded, after reduction, the diastereomeric mixture of syn- and anti-diaminophosphonates $4a$ and $4e-h$ in approximately the same ratio as starting nitro adducts 3 [\(Table 2,](#page-2-0) entries 2, 6–7, and 9–10, respectively). These results confirm that reduction of the nitro group

^{\dagger} When the solution of pure syn-3a in CDCl₃ was kept in the presence of pyridine for 4 h at room temperature, or when pure syn-3a was stirred in toluene in the presence of $K_2CO_3/TBAB$ system for 48 h at room temperature, followed by standard work-up, epimerized mixture of syn- and anti-3a (syn/anti=58/42) was always formed. No formation of starting diethyl nitromethanephosphonate (1) was observed in the ³¹P NMR spectrum of the reaction mixture, however.

1.2-Diaminophosphonates $4a-h$ prepared

^a Overall yield of pure syn- and *anti*-adducts $3a-h$ after flash chromatography. b Diastereomeric ratio measured by ³¹P NMR of the crude products.

^c The mixture of adducts **3a** (syn/anti=25/75), isolated from the mother liquor was used as substrate.

 d As a result of overlapping of $31P$ NMR signals, syn/anti ratio of diastereomeric diaminophosphonates 4 was determined by their cyclization into trans- and cisimidazolidine-2-thiones 5 (31 P NMR).

The mixture of nitro adducts 3g (syn/anti= \sim 10/90), isolated from the mother liquor, was used as substrate. As a result of relatively quick epimerization of 3g approximate ratio of diasteromer is only given.

takes place with retention of configuration on the carbon atom next to this group and simultaneously, correlate the configuration of 1,2-diaminophosphonates 4 with parent N-carbamate protected nitrophosphonates 3.

Relative configuration of 1,2-diaminophosphonates 4, in turn, was assigned by conversion of the selected adducts 4 into the appropriate imidazolidine-2-thiones 5 for which, relative configuration could easily be determined by $31P$ NMR analysis of their chemical shifts and the ring protons vicinal coupling constant ($^3\!J_{\rm HH})$ in ¹H NMR (Scheme 3).

As shown in Scheme 3, imidazolidine-2-thiones 5a and 5b were prepared in 51% and 41% overall yield via standard deprotection (TFA/CH2Cl2) of N-Boc protected syn-1,2-diamino-2-phenylethylphosphonate 4a and syn/anti-1,2-diaminobutylphosphonates 4f, followed by the conversion of free 1,2-diaminophosphonates thus obtained into trans-imidazolidine-2-thione 5a and the mixture of *trans-* and *cis-imidazolidine-2-thiones* **5b** (trans/cis= $64/36$), using 1,1'-thiocarbonyldiimidazole as condensing agent in the presence of

(syn/anti = ~10/90) $δ_P (trans/cis) = 18.46/16.42$

Scheme 3. Reagents and conditions: (i) TFA, CH_2Cl_2 , rt, 2 h; (ii) Et₃N, 1,1'-thiocarbonyldiimidazole, CH₂Cl₂, rt, 24 h; (iii) aq HCl; (iv) ammonium formate (10 equiv), 10% Pd/C (cat), MeOH, rt, 2 h, Ar.

triethylamine.²⁴ Similarly, the mixture of trans- and cis- $4g$ (cleavage of N-Cbz group was accomplished using ammonium formate/ $Pd-C$ system²⁵) afforded the mixture of the appropriate trans- and cisimidazolidine-2-thiones $5a$ (trans/cis=16/84) in 23% yield.

The stereochemistry of the imidazolidine-2-thiones 5 was assigned by 1 H and 31 P NMR analysis. The values of the vicinal coupling constants for **5a** $(^{3}$ J_{HH(trans)}=6.25 Hz, 3 J_{HH(cis)}=10.0 Hz) and **5b** (3 H_{H(trans)}=6.36 Hz) are consistent with the observation that trans-imidazolidine-2-thiones have smaller coupling constants than the corresponding cis-diastereomer.^{[26](#page-8-0)} In the ^{31}P NMR spectra of **5a** (δ_P (trans/cis)=18.46/16.42) and **5b** (δ_P (trans/cis)= 19.24/18.21) the signals of the trans-isomers appeared 1.03–2.04 ppm downfield relative to those of the cis-isomers.^{[21k](#page-8-0)} As all steps of the above mentioned transformations occur with the retention of configurations, these results confirm the assignment of relative configuration in the parent nitro adducts 3.

Having established the conditions of the reaction of diethyl nitromethanephosphonate (1) with N-carbamate protected α -aminoalkyl-p-tolylsulfones 2 we next attempted the asymmetric PTC reaction, 1^f , 17^f using cinchona alkaloid derived quaternary ammonium salts as catalysts. Such chiral PTC catalysts were successfully employed in the highly stereoselective aza-Henry reactions of simple nitroalkanes and N-protected imines generated in situ from N-formyl or N-carbamate α -aminoalkyl-arylsulfones.^{[5a,b,d](#page-7-0)}

N-tert-Butoxycarbonylamino(phenyl)methyl-p-tolylsulfone (2a) was chosen as a model compound, and its reaction with 1 in the presence of solid K_2CO_3 or Cs_2CO_3 was investigated using quininederived catalyst 6 (Scheme 4).

However, irrespective of the base applied, model studies did not give significant asymmetric induction under these conditions. The major syn-3a' was formed with low diastereoselectivity and poor enantioselectivity [\(Table 3,](#page-3-0) entries 1 and 2). The enantiomeric excesses were assessed from 31P NMR spectra of the crude reaction mixture, taking into account diagnostic chemical shifts (δ P) of diethyl syn- and anti-{(2-tert-butoxycarbonylamino)-1-[(S)-2-(6 methoxynaphthalen-2-yl)propionylamino]-2-phenyl-ethyl}phos-

Scheme 4. Asymmetric aza-Henry reaction of 1 with 2a.

phonates prepared by reduction of the crude nitro adducts **3a'** (NaBH₄/NiCl₂), followed by derivatization^{[27](#page-8-0)} of thus formed $syn/anti-4a'$ amines with (S)-Naproxen chloride $[(S)-2-(6-1)]$ methoxynaphthalen-2-yl)propionyl chloride].^{[28](#page-8-0)}

(S)-Naproxen chloride was proved to be a convenient chiral derivatizing agent (CDA) for the determination of enantiomeric purity as well as the absolute configuration assignments of diethyl 1-amino- and 2-aminoalkanephosphonates by NMR.^{27,29} Amidation under these conditions proceeds quantitatively, and is not accompanied by racemization.^{[27](#page-8-0) \ddagger} The diastereomeric anisochronicity in naproxen amide of $4a'$ was sufficiently large to estimate the ee of the syn-isomer ($\delta p=21.63/22.06$, $\Delta \delta = 0.43$). However, for the anti-isomer only a broad singlet ($\delta_{\rm P}$ =22.38) was observed in ³¹P NMR spectrum.

Entire derivatization (scale: 0.05 mmol) using (S)-Naproxen chloride as CDA, followed by determination of the ee as well as the absolute configurations assignments were performed in an NMR tube using anhydrous $CDCl₃$ as solvent and pyridine as a base.

^a Reaction conditions: scale 0.1 mmol, (c 0.1 mol/L); cat 6 (10 mol %) or cat 7 (20 mol %), base (5 equiv). Isomers 3a', 4a', and 3g' have the same structure as their racemic counterparts.

b Diastereomeric ratio measured by ³¹P NMR of the crude products.

Calculated from the ³¹P NMR spectra of (S)-Naproxen amides of syn/anti-4a' amines (δ_{syn} =21.63/22.06; δ_{ant} =22.38) obtained via reduction of 3a' using NaBH₄/NiCl₂ system, followed by derivatization of thus formed amines 4a' with (S)-Nap-Cl. Absolute configurations of the major syn-isomers 3a' and 4a', respectively, were determined to be (1S,2S) For the reaction of 1 with N-Cbz sulfone 2g the mixture of syn/anti-3g' (syn/anti=58/42) was obtained with 34% ee for syn-3g' under the same reaction conditions. In turn, only 68% conversion was obtained for the reaction conducted at 0 \degree C for 3 days.

When the reaction of 1 with N-Boc sulfone 2a was repeated at rt on 1 mmol scale, pure syn-nitro adduct 3a' and syn-diamino adduct 4a' (47% ee) were isolated in 27% and 61% yields, respectively.

^f When the reaction of 1 with N-Boc imine, preformed independently from 2a, was repeated at room temperature without the addition of K₂CO₃, only in the presence of catalyst **7**, 68% conversion was obtained after 24 h (syn/anti-**3a**′=56/44) and no syn-enantioselectivity was observed.
^g After 24 h only 10% conversion was observed.

Much better enantioselectivities were obtained when PTC catalyst 6 was substituted for the bifunctional thiourea derived from 9-amino(9-deoxy)epiquinine^{[30](#page-8-0)} 7 in the model reaction. [\(Scheme 4,](#page-2-0) Table 3, entries $3-8$). Recently, thiourea-based bifunctional organocatalysts were successfully employed in nitro-Mannich reactions of nitroalkanes with activated imines ^{[4a,b,h,k](#page-7-0)} Thus, in the reaction of 1a with 2a, in the presence of $7(20 \text{ mol }\%)$, the mixture of syn-3a' and $anti-3a'$ was formed with low syn-diastereoselectivity and with moderate to good syn-enantioselectivity, the latter being highly solvent and temperature dependent. The use of toluene gave the highest enantioselectivity at room temperature (47% ee, Table 3, entry 6). Lower enantioselectivities were observed for other solvents (Table 3, entries $3-5$). By lowering the reaction temperature to -20 °C, the ee was improved up to 67% for the major syn-3a', however at the expense of the reaction time (Table 3, entry 8). We also found that substitution of α -amido sulfone 2a for N-benzyloxycarbonylamino(phenyl)methyl-p-tolylsulfone (2g) resulted in the formation of a mixture of syn/anti-3g' (syn/anti=58/42) with lower syn-enantioselectivity (34% ee) compared with the analogous reaction in which sulfone 2a was used as imine precursor (Table 3, footnote d). However, when the addition of diethyl nitromethanephosphonate (1) to N-Boc imine,^{16c} preformed independently from N-Boc-amino(phenyl)methyl-p-tolylsulfone (2a), was repeated at room temperature in toluene without the addition of K_2CO_3 , only in the presence of chiral catalyst 7, the reaction was sluggish and gave a mixture of syn/anti-3a' (syn/anti=56/44) with no syn-enantioselectivity (Table 3, footnote f). This result shows that the catalyst 7 could only act as the chiral Brønsted acid in the reactions accomplished in the presence of K_2CO_3 .

The results obtained for the model reaction are encouraging, however, to achieve better enantioselectivity further tuning of the catalyst structure is necessary.

Finally, the absolute configuration of two consecutive stereogenic centers in the enantioenriched syn- $4a'$ was determined by $31P$ NMR analysis using (S) -Naproxen chloride as chiral derivatizing agent[§] for amidation 29,31 of each stereogenic center separately (Scheme 5). As shown in Scheme 5, nitro adducts $syn-3a'$ were reduced to partially protected syn- $4a'$ amines, followed by derivatization with (S)-Naproxen chloride²⁷ to give the mixture of diasteromeric (S)-Naproxen amides syn-8a' ($\delta_{\rm P}$ =22.06 (major) and 21.63 (minor); ratio: 74/26). According to the model proposed by us^{29} for the Naproxen amides of 1-amino- and 2-aminoalkanephosphonates ([Scheme 6](#page-4-0)), the upfield $31P$ $31P$ NMR chemical shift, attributed to the shielding effect of the naproxen naphthyl ring on the phosphorus atom, should be assigned to aminophosphonate with configuration (1S) at C-1 stereogenic center, whereas a minor, downfield chemical shift corresponds to configuration $(1R)$ at the same center.

In turn, bisamides syn-9a' (δ p=21.61 (minor), 21.40 (major); ratio: $26/74$) were prepared via consecutive acylation of syn- $4a'$ using acetic anhydride as acetylating agent, chemoselective deprotection of Boc group in thus obtained N-acetyl derivatives followed by derivatization

Scheme 5. The absolute configuration determination of the enantioenriched syn- $4a'$ amines by $31P$ NMR spectroscopy. Reagents and conditions: (i) NaBH₄ (10 equiv), NiCl₂×6H₂O (1.05 equiv), MeOH, -30 °C, 5 min, 0 °C; (ii) aq NH₄Cl, 0 °C; (iii) (S)-Nap-Cl (2.5 equiv), Pyridine/CDCl₃, rt, 1 h; (iv) Ac₂O (3 equiv), Et₃N (3 equiv), 4 h, rt; (v) TFA/ CH_2Cl_2 (1:1 v/v), rt, 2 h.

of free amino group using (S)-Naproxen chloride. Employing the model mentioned above,^{[29](#page-8-0)} [\(Scheme 6\)](#page-4-0) the major, upfield $31P$ NMR chemical shift should be assigned to aminophosphonate with configuration (2S) at carbon C-2, whereas downfield one corresponds to configuration (2R) at the same center.^{\P} Thus, the absolute configurations of major stereoisomers of $syn-**8a**'$ and $syn-**9a**'$ were determined to be (1S) and (2S), respectively. This allows assigning the same configuration for parent syn- $4a'$ and syn- $3a'$.

Since the diaminoderivative syn - $4a$ ['] is a mixture of enantiomers it is enough to use one enantiomer of Naproxen chloride for the absolute configuration assignment.

[{] As a matter of fact, the configuration at C-2 carbon atom could also be assigned based on the absolute configuration at C-1 stereogenic center in $syn-8a'$ and relative configuration of compounds 3 and 4.

Scheme 6. Model for the assignment of the configuration of diethyl 1-amino- and 2 aminoalkylphosphonates from the ³¹P NMR spectra of their (S)-Naproxen amides, based on the shielding effect exerted by the anisotropy cone of the naproxen naphthyl ring on the substituents in aminophosphonates 29 .

3. Conclusions

In conclusion, we have shown that the reaction of diethyl nitromethanephosphonate (1) with N-Boc and N-Cbz protected α -aminoalkyl-p-tolylsulfones **2a–h**, as stable equivalents of N-carbamate imines, is a convenient and efficient method for the preparation of N-protected β -amino- α -nitrophosphonates 3g-h and in consequence also α , β -diaminophosphonates **3a**-h. Additions are accomplished under PTC conditions and give the mixture of synand *anti*- nitro adducts $3g-h$ in high yields and low syn-diastereoselectivity. Homologue of the nitrophosphonate 1 is unreactive under these conditions. Crystallization of the reaction mixture allows obtaining pure syn-3a-d and syn-3g derived from the corresponding aromatic aldehydes, which, after the reduction by NaBH₄/NiCl₂ system afford partially protected syn-1,2-diaminoalkanephosphonates $4a-d$ and $4g$ in moderate yields. A model catalytic, enantioselective version of this aza-Henry reaction was only partially successful. The moderate enantioselectivity (67%) achieved for syn-diastereomer $\mathbf{3a}'$, when bifunctional thiourea derived from 9-amino(9-deoxy)epiquinine 7 is used as a catalyst in the presence of K_2CO_3 . Using $31P$ NMR spectroscopy and (S)-Naproxen chloride as chiral derivatizing agent the absolute configuration of the consecutive stereogenic centers in $syn-4a'$ amine was determined.

4. Experimental

4.1. General

NMR spectra were recorded on a Bruker Avance DPX 250 instrument at 250.13 MHz for $^1\mathrm{H}$ NMR, 62.90 MHz for 13 C NMR, and 101.3 MHz for $31P$ NMR in CDCl₃ solution, using either tetramethylsilane as an internal and 85% H₃PO₄ as an external standard. Positive chemical shifts are downfield from external 85% H₃PO₄ for $31P$ NMR spectra. Chemical shifts (δ) are indicated in parts per million and coupling constants (*J*) in hertz. For ¹³C NMR spectra, the peak assignments were made with the assistance of CH-COSY and DEPT experiments. Partially overlapped signals are assigned by asterisks (*). The enantiomeric excesses (ee) were determined by $31P$ NMR spectroscopy using (S)-Naproxen chloride as chiral derivatizing agent (CDA). Optical rotations were measured on a PolAAr 3001 (Optical Activity Ltd.) polarimeter. Elemental analyses were performed on a Perkin-Elmer PE 2400 Analyzer. IR spectra were measured on an IR Alpha Bruker (ATR) instrument and are reported in cm⁻¹. Melting points were determined in open capillaries and

are uncorrected. All reagents were purchased from Fluka and were used without further purification. Diethyl nitromethanephosphonate,¹³ diethyl 1-nitroethanephosphonate,¹⁸ and N -Boc α -amidoalkyl-p-tolylsulfones^{[15](#page-8-0)} were prepared as described previously.

4.2. Preparation of substituted diethyl N-carbamate 2-amino-1-nitroalkanephosphonates 3a-h; general procedure

Solid, anhydrous K_2CO_3 (1.40 g, 10.0 mmol) was added to a solution of diethyl nitromethanephosphonate (1, 394 mg, 2.0 mmol), α -amido sulfone (2, 2.05 mmol) and tetrabutylammonium bromide (TBAB, 70 mg, 0.20 mmol) in anhydrous toluene (20 mL). The suspension was vigorously stirred for 26 h at room temperature. The mixture was carefully quenched with 10% aq KHSO₄ (10 mL), the organic layer separated and the aqueous phase extracted with $CH₂Cl₂ (3×20 mL)$. The combined organic phases were washed with brine (2×10 mL), dried over MgSO₄, evaporated, and subjected to flash chromatography on silica gel (AcOEt/hexanes 5:2 or 5:3 v/v) to give pure mixtures of syn- and anti-adducts $3a-h$. Pure syn- $3a-d$ and syn-3g were obtained after crystallization from $CCl₄$ [\(Table 1](#page-1-0)).

4.2.1. Diethyl syn-(2-tert-butoxycarbonylamino-1-nitro-2-phenylethyl) phosphonate 3a. Crude product was purified by flash chromatography (AcOEt/hexanes 5:2 v/v), followed by crystallization from CCl₄ to give the title compound 3a (322 mg, 40%) as a colorless solid, mp 136-139 °C; [found: C, 50.73; H, 6.82; N, 6.98. C₁₇H₂₇N₂O₇P requires C, 50.74; H, 6.76; N, 6.96%]; Rf (AcOEt/hexanes 5:2 v/v) 0.44; v_{max} (ATR) 3315, 2974, 1711, 1560, 1520, 1256, 1161, 982, 699; δ_P (101 MHz, CDCl₃) 10.36; δ_H (250 MHz, CDCl₃) 7.38–7.29 (m, 5H_{Ar}), 6.09–5.95 (m, 1H, NHBoc), 5.77-5.50 (m, 1H, CHNHBoc), 5.34, 5.28 (dd and br dd, respectively, 1H, 3 J_{HH} 7.4, 5.8 Hz, 2 J_{HP} 13.0, 16.0 Hz, CHNO₂, rotamers), 4.35-3.80 (m, 4H, 2CH₃CH₂O), 1.42, 1.41 (2s, 9H, CH_3 ₃C, rotamers), 1.36-1.10 (m, 6H, 2CH₃CH₂O); δ_C (63 MHz, CDCl₃) 154.4 (s, C=O), 136.8 (d, 3 J_{CP} 6.4 Hz, CHC_{Ar}), 128.6, 128.3, 126.8 (s, 5CH_{Ar}), 88.4 (d, ¹J_{CP} 141.1 Hz, CHNO₂), 80.1 (s, (CH₃)₃C), 64.2 (d, ²J_{CP} 6.1 Hz, 2CH₃CH₂O), 53.9 (s, CHNHBoc), 28.1 (s, $(CH_3)_3C$), 16.0, 15.9 (2d, $^3J_{CP}$ 6.0, 5.8 Hz, $2CH_3CH_2O$).

4.2.2. Diethyl syn-[2-tert-butoxycarbonylamino-2-(4-methoxyphenyl)- 1-nitroethyllphosphonate 3b. Crude product was purified by flash chromatography (AcOEt/hexanes 5:2 v/v), followed by crystallization from CCl₄ to give the title compound **3b** (337 mg, 39%) as a colorless solid, mp 145–148 °C; [found: C, 49.98; H, 6.63; N, 6.33. C₁₈H₂₉N₂O₈P requires C, 50.00; H, 6.76; N, 6.48%]; R_f (AcOEt/hexanes 5:2 v/v) 0.39; v_{max} (ATR) 3334, 2939, 1704, 1559, 1517, 1247, 1160, 1012, 841, 714; δ_{P} (101 MHz, CDCl₃) 10.08; δ_H (250 MHz, CDCl₃) 7.31–6.85 (m, 4H_{Ar}), 6.17 (d, 1H, 3 J_{HH} 9.2 Hz, NHBoc), 5.55–5.42 (m, 1H, CHNHBoc), 5.48-5.29 (m, 1H, CHNO₂), 4.33-3.96 (m, 4H, 2CH₃CH₂O), 3.78 (s, 3H, OCH₃), 1.41 (s, 9H, (CH₃)₃C), 1.39-1.16 (m, 6H, 2CH₃CH₂O); δ_C (63 MHz, CDCl₃) 157.8 (s, C=O), 152.8 (s, CH₃OC_{Ar}), 127.1 (d, ³J_{CP} 6.2 Hz, CHC_{Ar}), 112.4, 126.5, (2s, 4CH_{Ar}), 86.8 (d, ¹J_{CP} 141.1 Hz, CHNO₂), 78.4 (s, (CH₃)₃C), 62.5 (d, ²J_{CP} 6.4 Hz, 2CH₃CH₂O), 53.4 (s, OCH₃), 52.1 (br s, CHNHBoc), 26.4 (s, $(CH_3)_3C$), 14.4, 14.3 (2d, $3J_{CP}$ 5.6 Hz, 2CH₃CH₂O).

4.2.3. Diethyl syn-[2-tert-butoxycarbonylamino-2-(4-chlorophenyl)- 1-nitroethyllphosphonate $3c$. Crude product was purified by flash chromatography (AcOEt/hexanes 5:3 v/v), followed by crystallization from CCl₄ to give the *title compound* **3c** (358 mg, 41%) as a colorless solid, mp 157-160 °C; [found: C, 46.52; H, 5.85; N, 6.65. $C_{17}H_{26}CIN_2O_7P$ requires C, 46.74; H, 6.00; N, 6.41%]; R_f (AcOEt/hexanes 5:2 v/v) 0.39; v_{max} (ATR) 3332, 2974, 1709, 1560, 1514, 1246, 1159, 1011, 867; δ_P (101 MHz, CDCl₃) 10.10; δ_H (250 MHz, CDCl₃) 7.35–6.85 (m, 4H_{Ar}), 6.09 (br d, 3 J_{HH} 8.1 Hz, 1H, NHBoc), 5.60–5.48 (m, 1H_i CHNHBoc), 5.31 (br dd, $^{3}J_{HH}$ 7.8 Hz, $^{2}J_{HP}$ 13.0 Hz, 1H, CHNO₂),

4.37-3.92 (m, 4H, 2CH₃CH₂O), 1.41 (s, 9H, (CH₃)₃C), 1.38-1.19 (m, 6H, 2CH3CH2O); $\delta_{\mathsf C}$ (63 MHz, CDCl3) 152.8 (s, C=O), 133.8 (d, 3 J_{CP} 6.5 Hz, CHC_{Ar}), 132.2 (C_{Ar}), 126.7, 126.3 (CH_{Ar}), 86.4 (d, ¹J_{CP} 141.0 Hz, CHNO₂), 78.7 (s, (CH₃)₃C), 62.9, 63.8 (2d, ²J_{CP} 6.3, 7.5 Hz, 2CH₃CH₂O), 51.7 (br s, CHNHBoc), 26.4 (s, (CH₃)₃C), 14.4, 14.3 (2d, ³J_{CP} 6.0 Hz, 2CH₃CH₂O).

4.2.4. Diethyl syn-(2-tert-butoxycarbonylamino-2-furan-2-yl-1-nitroethyl)phosphonate 3d. Crude product was purified by flash chromatography (AcOEt/hexanes 5:2 v/v), followed by crystallization from CCl₄ to give the *title compound* **3d** (322 mg, 41%) as a colorless solid, mp $92-95$ °C; [found: C, 45.86; H, 6.42; N, 7.03. $C_{15}H_{25}N_2O_8P$ requires C, 45.92; H, 6.42; N, 7.14%]; R_f (AcOEt/hexanes 5:2 v/v) 0.44; v_{max} (ATR) 3294, 2976, 1714, 1555, 1390, 1270, 1245, 1022, 752; δ_P (101 MHz, CDCl₃) 9.88; δ_H (250 MHz, CDCl₃) 7.38–7.36 (m, 1H, H_{Ar}), 6.37–6.32 (m, 2H_{Ar}), 5.83–5.62 (m, 2H, CHNHBoc, NHBoc), 5.50–5.39 (m, 1H, CHNO₂), 4.33–3.95 (m, 4H, 2CH₃CH₂O), 1.45, 1.44 (2s, 9H, $(CH_3)_3C$; rotamers), 1.38–1.18 (m, 6H, 2CH₃CH₂O); $\delta_{\sf C}$ (63 MHz, CDCl₃) 152.7 (s, C=0), 147.5 (d, 3 J_{CP} 7.3 Hz, CHC_{Ar}), 141.0, 108.9, 106.7 (CH_{Ar}), 84.2 (d, ¹J_{CP} 142.0 Hz, CHNO₂), 78.7 (s, (CH₃)₃C), 63.1, 62.6 (2d, 2 J_{CP} 6.5, 7.0 Hz, 2CH₃CH₂O), 46.8 (s, CHNHBoc), 26.4 (s, (CH₃)3C), 14.5, 14.2 (2d, 3 J_{CP} 5.7 Hz, 2CH₃CH₂O).

4.2.5. Diethyl syn/anti-(2-tert-butoxycarbonylamino-1-nitropropyl) phosphonates 3e. Crude product was purified by flash chromatography (AcOEt/hexanes 5:3 v/v) to give the title compound 3e (599 mg, 88%) as a colorless viscous oil; [found: C, 42.55; H, 7.60; N, 8.08. $C_{12}H_{25}N_2O_7P$ requires C, 42.35; H, 7.40; N, 8.23]; R_f (AcOEt/hexanes 5:3 v/v) 0.37; v_{max} (neat) 3305, 2980, 1701, 1556, 1249, 1160, 1013, 730; δ_P (101 MHz, CDCl₃) 11.27 (anti-3e 39%), 11.09 (syn-3e 61%); δ_H (250 MHz, CDCl₃) 5.34–5.00 (m, 2H, CHNO₂, NHBoc, syn and anti), 4.54 -4.36 (m, 1H, CHNHBoc, syn and anti), 4.33 -4.08 (m, 4H, 2CH₃CH₂O, syn and anti), 1.45-1.33 (m, 18H, 2CH₃CH₂O, CH₃CH, $(CH_3)_3C$, syn and anti); δ_C (63 MHz, CDCl₃) 152.9 (br s, C=0, syn and anti), 86.1, 84.8 (2d, $^{1}\!J_{\rm CP}$ 143.3, 141.3 Hz, CHNO $_2$, syn and anti), 78.3 (br s, (CH₃)₃C, syn and anti), 62.9, 62.7, 62.5, 62.2 (4d, ²J_{CP} 6.7, 7.0, 7.9, 6.8 Hz, $2CH_3CH_2O$, syn and anti), 45.4 (s, CHNHBoc, syn), 45.2 (s, CHNHBoc, anti), 26.4 (s, $(CH_3)_3C$, syn and anti), 16.5 (s, CH₃CH, syn), 15.5 (s, CH₃CH, anti), 14.4 (d, 3 J_{CP} 5.8 Hz, 2CH₃CH₂O, syn and anti).

4.2.6. Diethyl syn/anti-(2-tert-butoxycarbonylamino-1-nitrobutyl) phosphonates 3f. Crude product was purified by flash chromatography (AcOEt/hexanes 5:3 v/v) to give the *title compound* 3f (560 mg, 79%) as a colorless viscous oil; [found: C, 44.27; H, 7.72; N, 7.71. $C_{13}H_{27}N_2O_7P$ requires C, 44.07; H, 7.68; N, 7.91%]; R_f (AcOEt/hexanes 5:3 v/v) 0.39; v_{max} (neat) 2975, 1716, 1557, 1315, 1244, 1162, 1010, 730, 567, 464; δ_P (101 MHz, CDCl₃) 11.29 (anti-3f 44%), 11.0 (syn-3f 56%); δ_H (250 MHz, CDCl₃) 5.25-5.09 (m, 2H, CHNO₂, NHBoc, syn and anti), 4.36-4.16 (m, 5H, 2CH₃CH₂O, CHNHBoc, syn and anti), 1.80-1.60 (m, 2H, CH₃CH₂CH, syn and anti), 1.44 (s, 9H, $(CH₃)₃C$, syn and anti), 1.41–1.31 (m, 6H, 2CH₃CH₂O, syn and anti), 1.00 (br t, 3 J_{HH} 7.4 Hz, 3H, CH₃CH₂CH, syn and anti); δ_C (63 MHz, CDCl₃) 155.1, 154.9 (2s, C=O, syn and *anti*), 86.8 (d, ¹J_{CP} 142.4 Hz, CHNO₂, syn), 86.3 (d, ¹J_{CP} 145.6 Hz, CHNO₂, anti), 79.8 (br s, (CH₃)₃C, syn and anti), 64.9, 64.4, 64.3, 63.8 (4d, 2 J_{CP} 6.8, 5.0, 4.0, 6.9, 2CH₃CH₂O, syn and *anti*), 52.9 (s, CHNHBoc, anti), 52.1 (s, CHNHBoc, syn), 28.1 (s, (CH₃)3C, syn and anti), 25.3 (d, $^3\!J_{\rm CP}$ 5.2 Hz, CH₃CH₂, syn), 24.7 (d, 3 J_{CP} 4.3 Hz, CH₃CH₂, anti), 16.1 (d, 3 J_{CP} 6.0 Hz, 2CH₃CH₂O, syn and *anti*), 10.3 (d, 4 J_{CP} 2.9 Hz, CH₃CH₂).

4.2.7. Diethyl syn-(2-benzyloxycarbonylamino-1-nitro-2-phenylethyl) phosphonate 3g. Crude product was purified by flash chromatography (AcOEt/hexanes 5:2 v/v), followed by crystallization from CCl₄. to give the title compound 3g (244 mg, 28%) as a colorless solid, mp 100-103 °C; [found: C, 55.16; H, 5.60; N, 6.38. C₂₀H₂₅N₂O₇P requires C, 55.05; H, 5.77; N, 6.42%]; R_f (AcOEt/hexanes 5:2 v/v) 0.39; ν_{max} (ATR) 3306, 2957, 1742, 1555, 1528, 1246, 1006, 813, 694; δ_P (101 MHz, CDCl₃) 10.93 (anti-3g 47%), 10.17 (syn-3g 53%); δ_H (250 MHz, CDCl₃)

7.39–7.29 (m, 10H_{Ar}), 6.40 (br d, 3 J_{HH} 9.3 Hz, 1H, NH), 5.70–5.62 (m, 1H, CHPh), 5.40 (dd, 3 J_{HH} 6.1 Hz, 2 J_{HP} 16.5 Hz, 1H, CHNO₂), 5.17–5.03 (m, 2H, CH₂Ph), 4.15-4.01 (m, 4H, 2CH₃CH₂O), 1.33-1.08 (m, 6H, 2CH₃CH₂O); δ_C (63 MHz, CDCl₃) 155.3 (s, C=O), 136.3 (d, ³J_{CP} 5.8 Hz, CHC_{Ar}), 136.0 (s, C_{Ar}), 128.7, 128.5, 128.3, 128.0, 127.2 (s, CH_{Ar}), 88.1 (d, $J_{\rm CP}$ 141.0 Hz, CHNO $_2$), 67.0 (s, CH $_2$ Ph), 64.3 (d, 2 J $_{\rm CP}$ 6.0 Hz, 2CH $_3$ CH $_2$ O), 54.6 (s, CHNH), 16.0, 15.9 (2d, 3 J_{CP} 6.3 Hz, 2CH₃CH₂O).

4.2.8. Diethyl syn/anti-(2-benzyloxycarbonylamino-1-nitropropyl) phosphonates 3h. Crude product was purified by flash chromatography (AcOEt/hexanes 5:3 v/v) to give the title compound 3h (623 mg, 84%) as a colorless viscous oil; [found: C, 48.33; H, 6.37; N, 7.21. C₁₅H₂₃N₂O₇P requires C, 48.13; H, 6.19; N, 7.48%]; R_f (AcOEt/hexanes 5:2 v/v) 0.43; v_{max} (neat) 3285, 2983, 1718, 1555, 1245, 1012, 697, 671; δ_P (101 MHz, CDCl₃) 10.81 (anti-**3h** 43%), 10.67 (syn-**3h** 57%); ô_H (250 MHz, CDCl₃) 7.36–7.29 (m, 5H_{Ar}), 5.93 (d, 1H,
³J_{HH} 8.9 Hz, NHCbz, syn), 5.84 (d, ³J_{HH} 8.3 Hz, NHCbz, 1H, anti), 5.26 (dd, ³J_{HH} 5.2 Hz, ²J_{HP} 14.5 Hz, 1H, CHNO₂, anti), 5.16 (dd, ³J_{HH} 7.2 Hz, ²L_{im} 15.2 Hz, 2H CH₂Ph, 5.2H CH₂Ph, 5.2H CH₂Ph 2 J_{HP} 15.2 Hz, 1H, CHNO₂, syn), 5.10, 5.05 (2s, 2H, CH₂Ph, syn and anti), 4.67-4.39 (m, 1H, CHNHCbz, syn and anti), 4.25-4.06 (m, 4H, $2CH_3CH_2O$, syn and anti), 1.42, 1.39 (2d, $3J_{HH}$ 3.0 Hz, 3H, CH₃CH, syn and anti), 1.36–1.26 (m, 6H, 2CH₃CH₂O, syn and anti); δ _C (63 MHz, CDCl₃) 153.5 (s, C=O, syn and anti), 134.4, 134.3, (2s, CH₂C_{Ar}, syn and anti), 126.7, 126.4, 126.3 (3s, CH_{Ar}, syn and anti), 86.0, 84.8 (2d, ¹J_{CF} 145.0, 141.9 Hz, CHNO₂, syn and anti), 65.1 (s, CH₂Ph, syn and anti), 63.0, 62.7, 62.6, 62.2 (4d, 2 J_{CP} 6.7, 6.8, 7.5, 6.8 Hz, 2CH₃CH₂O, syn and anti), 45.9, 45.7 (2s, CHNH, syn and anti), 16.5 (s, CH₃CH, syn), 15.4 (s, CH₃CH, anti), 14.4 (d, 3 J_{CP} 5.7 Hz, 2CH₃CH₂O, syn and anti).

4.3. Preparation of diethyl 2-(N-carbamate) protected 1,2 diaminoalkanephosphonates 4a-h; general procedure

Sodium borohydride (113 mg, 3.0 mmol) was added in two portions to a cooled to -30 °C solution of diethyl N-carbamate protected 2-amino-1-nitroalkanephosphonate 3 (0.3 mmol) and $NiCl₂×6H₂O$ (71 mg, 0.3 mmol) in methanol (3 mL). The mixture was allowed to warm up to 0 \degree C for 10 min, quenched with saturated aq NH₄Cl (9 mL), and extracted with CH_2Cl_2 (3×15 mL). The combined organic phases were washed with brine (5 mL), dried over MgSO4, evaporated, and subjected to flash chromatography on silica gel (AcOEt) to give pure adducts 4.

4.3.1. Diethyl syn-(1-amino-2-tert-butoxycarbonylamino-2-phenylethyl)phosphonate 4a. Crude product was purified by flash chromatography (AcOEt) to give the title compound 4a (70 mg, 63%) as a colorless solid, mp 108-111 °C; [found: C, 54.63; H, 7.60; N, 7.30. C₁₇H₂₉N₂O₅P requires C, 54.83; H, 7.85; N, 7.52%]; R_f (AcOEt) 0.26; v_{max} (ATR) 3391, 3032, 1703, 1530, 1362, 1265, 1246, 1212, 1017, 966, 933, 860, 703; δ_P (101 MHz, CDCl₃) 25.69; δ_H (250 MHz, CDCl₃) 7.42–7.22 (m, 5H_{Ar}), 6.06 (brs, 1H, NHBoc), 5.06 (ddd, ³J_{HH} 3.3, 8.0 Hz, 3 J_{HP} 11.6 Hz, 1H, CHNHBoc), 4.22–3.99 (m, 4H, 2CH₃CH₂O), 3.40 (dd, 3 J_{HH} 3.3 Hz, 2 J_{HP} 15.6 Hz, 1H, CHNH₂), 1.66 (br s, 2H₁ CHNH₂), 1.40 (s, 9H, (CH₃)₃C), 1.35, 1.25 (2 t, ³J_{HH} 7.1 Hz, 6H, 2CH₃CH₂O); δ_C (63 MHz, CDCl₃) 155.2 (s, C=O), 140.3 (d, ³J_{CP} 4.7 Hz, CHC_{Ar}), 128.5, 127.4, 126.6 (C_{ar}), 79.4 (s, (CH₃)₃C), 62.6, 62.4 (2d, ²J_{CF} 6.9 Hz, 2CH₃CH₂O), 54.9 (br s, CHNHBoc), 53.5 (d, $^{1}J_{CP}$ 152.0 Hz, CHNH₂), 28.3 (s, (CH₃)₃C), 16.5, 16.3 (2d, ³J_{CP} 6.7, 7.0 Hz, 2CH₃CH₂O).

4.3.2. Diethyl syn-[1-amino-2-tert-butoxycarbonylamino-2-(4-methoxyphenyl)-ethyllphosphonate $4b$. Crude product was purified by flash chromatography (AcOEt) to give the title compound 4b (64 mg, 53%) as a colorless solid, mp 96-99 °C; [found: C, 53.51; H, 7.82; N, 6.92. C₁₈H₃₁N₂O₆P requires C, 53.72; H, 7.76; N, 6.96%]; R_f (AcOEt) 0.20; ν_{max} (ATR) 3276, 2977, 1704, 1511, 1244, 1158, 1026, 961, 767; δ_P (101 MHz, CDCl₃) 25.17; δ_H (250 MHz, CDCl₃) 7.29-7.22 (m, 2H_{Ar}), 6.90-6.84 (m, $2H_{Ar}$), 5.95 (br s, 1H, NHBoc), 4.95 (ddd, $^3J_{HH}$ 3.4, 8.0 Hz, $^3J_{HP}$ 13.4 Hz, 1H,

CHNHBoc), 4.21-3.98 (m, 4H, 2CH₃CH₂O), 3.79 (s, 3H, CH₃O), 3.36 (dd, $^3\!J_{\rm HH}$ 3.4 Hz, $^2\!J_{\rm HP}$ 15.0 Hz, 1H, CHNH₂), 1.45 (br s, 2H, NH₂), 1.39 (s, 9H, (CH₃)3C), 1.34, 1.27 (2 t, 3 J_{HH} 7.1 Hz, 6H, 2CH3CH2O), δ_{C} (63 MHz, CDCl₃) 158.8 (s, CH₃OC_{Ar}), 155.1 (s, C=O), 132.2 (br s, CHC_{Ar}), 127.6, 113.7 (CH_{Ar}), 79.2 (s, (CH₃)₃C), 62.4, 62.2 (2d, ²J_{CP} 6.9 Hz, 2CH₃CH₂O), 55.1 (s, CH₃O), 54.3 (br s, CHNHBoc), 53.5 (d, 1 J_{CP} 151.0 Hz, CHNH₂), 28.2 (s, (CH₃)₃C), 16.3, 16.2 (2d, $\frac{3}{2}$ _{CP} 6.6 Hz, 2CH₃CH₂O).

4.3.3. Diethyl syn-[1-amino-2-tert-butoxycarbonylamino-2-(4-chlorophenyl)-ethyl]phosphonate $4c$. Crude product was purified by flash chromatography (AcOEt) to give the title compound $4c$ (81 mg, 66%) as a colorless solid, mp $111-114$ °C; [found: C, 50.38; H, 7.04; N, 6.97. C₁₇H₂₈ClN₂O₅P requires C, 50.19; H, 6.94; N, 6.89%]; R_f (AcOEt) 0.26; v_{max} (ATR) 3273, 2931, 1703, 1532, 1247, 1212, 1161, 1012, 966, 738; δ_P (101 MHz, CDCl₃) 25.77; δ_H (250 MHz, CDCl₃) 7.32-7.12 (m, 4H_{Ar}), 6.20 (br s, 1H, NHBoc), 4.94 (ddd, 3 J_{HH} 3.4, 7.7 Hz, 3 J_{HP} 11.6 Hz, 1H, CHNHBoc), 4.13–3.93 (m, 4H, 2CH₃CH₂O), 3.26 (dd, ³J_{HH} 3.4 Hz, ²J_{HP} 16.1 Hz, 1H, CHNH₂), 1.50 (br s, 2H, NH₂), 1.32 (s, 9H, (CH₃)₃C), 1.34, 1.26 (2t, ³J_{HH} 7.1 Hz, 6H, CH₃CH₂O); δ_{C} (63 MHz, CDCl₃) 153.4 (s, C=O), 137.2 (br d, $^3\text{J}_\text{CP}$ 7.0 Hz, CHCAr), 131.3 (s, ClCAr), 126.7, 126.3 (2s, 2CHAr), 77.7 (s, (CH₃)₃C), 60.8, 60.6 (2d, ²J_{CP} 7.0 Hz, 2CH₃CH₂O), 52.7 (br s, CHNHBoc), 51.5 (d, 1 J_{CP} 152.1 Hz, CHNH₂), 26.5 (s, (CH₃)₃C), 14.6, 14.5 (2d, 3 J_{CP} 6.4 Hz, 2CH₃CH₂O).

4.3.4. Diethyl syn-[(1-amino-2-tert-butoxycarbonylamino-2-furan-2-yl)ethyl]phosphonate **4d**. Crude product was purified by flash chromatography (AcOEt) to give the title compound 4d (60 mg, 55%) as a colorless solid, mp 49-52 °C; [found: C, 49.53; H, 7.59; N, 7.80. $C_{15}H_{27}N_2O_6P$ requires C, 49.72; H, 7.51; N, 7.73%]; R_f (AcOEt) 0.26; v_{max} (ATR) 3311, 2980, 1703, 1526, 1215, 1161, 1018, 953, 759; δ_{P} (101 MHz, CDCl₃) 25.27; δ_H (250 MHz, CDCl₃) 7.36–7.35 (m, 1H_{Ar}), 6.33–6.31 (m, 1H_{Ar}), 6.27–6.25 (m, 1H_{Ar}), 5.95 (d, ³J_{HH} 8.8 Hz, 1H, NHBoc), 5.12 (ddd, $^3\!J_{\rm HH}$ 3.7, 8.8 Hz, $^3\!J_{\rm HP}$ 12.9 Hz, 1H, CHNHBoc), 4.23–3.93 (m, 4H, 2CH₃CH₂O), 3.53 (dd, ³J_{HH} 3.7 Hz, ²J_{HP} 15.6 Hz, 1H, CHNH₂), 1.69 (br s, 2H, NH₂), 1.44 (s, 9H, (CH₃)₃C), 1.33, 1.27 (2d, 3 J_{HH} 7.1 Hz, 6H, 2CH₃CH₂O); δ_{C} (63 MHz, CDCl₃) 154.9 (s, C=O), 153.0 (d, ${}^{3}J_{CP}$ 11.1 Hz, CHC_{Ar}), 141.2, 110.1, 106.9 (CH_{Ar}), 79.4 (s, (CH₃)3C), 62.3, 62.1 (2d, ²J_{CP} 7.0, 6.9 Hz, 2CH₃CH₂O), 51.5 (d, ¹J_{CP} 153.1 Hz, CHNH2), 50.1 (br s, CHNHBoc), 28.1 (s, (CH3)3C), 16.2, 16.1 (2d, $\rm{^3J_{CP}}$ 4.7 Hz, 2CH₃CH₂O).

4.3.5. Diethyl syn/anti-(1-amino-2-tert-butoxycarbonylamino-propyl)phosphonates **4e**. Crude product was purified by flash chromatography (AcOEt) to give the title compound 4e (40 mg, 43%) as a colorless oil; [found: C, 46.54; H, 8.54; N, 8.80. C₁₂H₂₇N₂O₅P requires C, 46.44; H, 8.77; N, 9.03%]; R_f (AcOEt) 0.29; v_{max} (neat) 3305, 2977, 1698, 1365, 1230, 1164, 1021, 959, 730; δ_P (101 MHz, CDCl₃) 27.16 (syn-4e 58%), 27.08 (anti-4e 43%); δ_H (250 MHz, CDCl₃) 5.24 (br s, 1H, NHBoc, syn and anti), $4.24-4.09$ (m, $4H$, $2CH_3CH_2O$, syn and *anti*), 4.08–3.83 (m, 1H, CHNHBoc, syn and *anti*), 3.17 (dd, 3 J_{HH} 3.4 Hz, 2 J_{HP} 15.5 Hz, 1H, CHNH₂, anti), 3.09 (dd, 3 J_{HH} 4.4 Hz, 2 J_{HP} 14.9 Hz, 1H, CHNH₂, syn), 1.77 (br s, 2H, CHNH₂, syn and anti), 1.44 (s, 9H, (CH₃)₃C, syn and anti), 1.35 (t, 3 J_{HH} 7.1 Hz, 6H, 2CH₃CH₂O, syn and *anti*), 1.26 (d, 3 J_{HH} 6.8 Hz, 3H, CH₃CH, syn), 1.22 (d, 3 J_{HH} 7.0 Hz, 3H, CH₃CH, anti); δ_C (63 MHz, CDCl₃) 153.4 (s, C=O, syn and anti), 77.1(s, $(CH_3)_3C$, syn and anti), 60.5 (br s, CHNHBoc, syn and anti), 60.4, 60.2 (2d, 2 J_{CP} 7.1, 7.0 Hz, 2CH₃CH₂O, syn and anti), 50.9 (d, ¹J_{CP} 148.3 Hz, CHNH₂, syn), 50.2 (d₁¹J_{CP} 144.1 Hz, CHNH₂, anti), 26.4 (s, (CH₃)₃C, syn and anti), 14.5 (d, 3 J_{CP} 5.6 Hz, 2CH₃CH₂O, syn and anti), 14.1 (br s, $CH₃CH$, syn and anti).

4.3.6. Diethyl syn/anti-(1-amino-2-tert-butoxycarbonylamino-butyl) phosphonates 4f. Crude product was purified by flash chromatography (AcOEt) to give the title compound 4f (58 mg, 60%) as a colorless oil; [found: C, 48.39; H, 9.26; N, 8.45. $C_{13}H_{29}N_2O_5P$ requires C, 48.14; H, 9.01; N, 8.64 %]; R_f (AcOEt) 0.30; ν_{max} (neat) 3300, 2975, 1706, 1507, 1238, 1164, 1021, 958, 783, 558; δ_P (101 MHz, CDCl₃) 27.5 (syn- and anti-4f); $\delta_{\rm H}$ (250 MHz, CDCl₃) 5.23, 5.12 (2br d, 3 J_{HH} 8.4, 9.4 Hz, 1H₁ CHNHBoc, syn and anti), 4.23-4.08 (m, 4H, 2CH₃CH₂O, syn and anti), 3.93–3.69 (m, 1H, CH₃CH₂CH, syn and anti), 3.17, 3.14 (2dd, 3 J_{HH} 3.6, 3.7 Hz $^2J_{HP}$ 15.6, 14.5 Hz, 1H, CHNH₂, syn and anti), 1.80–1.49 (m, 4H, NH_2 , CH₃CH₂CH, syn and anti), 1.44 (s, 9H, (CH₃)₃C, syn and anti), 1.35, 1.34 (2t, $3J_{HH}$ 7.1 Hz, 6H, 2CH₃CH₂O, syn and anti), 1.00–0.90 (m, 3H, CH₃CH₂CH, syn and anti); δ_c (63 MHz, CDCl₃) 153.9 (s, C=0, syn and anti), 77.4 $(s, (CH_3)_3C, syn$ and anti), 53.1, 52.2 (2br s, CHNHBoc, syn and anti), 60.7, 60.1 (2br d, 2 *J*_{CP} 7.9 Hz, 2CH₃CH₂O, syn and anti), 51.0, 50.6 (2d, 1 J_{CP} 144.9, 148.8 Hz, CHNH₂, syn and anti), 27.4 (s, (CH₃)₃C, syn and anti), 24.4, 23 (2br s, CH₃CH₂, syn and anti), 15.5 (d, 3 J_{CP} 5.3 Hz, $2CH_3CH_2O$, syn and anti), 9.9, 9.8 (2s CH_3CH_2 , syn and anti).

4.3.7. Diethyl syn-(1-amino-2-benzyloxycarbonylamino-2-phenylethyl)phosphonate $4g$. Crude product was purified by flash chromatography (AcOEt) to give the title compound **4g** (91 mg, 75%) as a colorless solid, mp $64-66$ °C; [found: C, 58.99; H, 6.66; N, 6.86. C20H27N2O5P requires C, 59.11; H, 6.70; N, 6.89%]; Rf (AcOEt) 0.20; v_{max} (ATR) 3267, 2980, 1716, 1534, 1249, 1017, 965, 790, 697, 527, 509; δ_P (101 MHz, CDCl₃) 25.48; δ_H (250 MHz, CDCl₃) 7.33–7.25 (m, 10H_{Ar}), 6.65 (br d, ${}^{3}J_{HH}$ 8.3 Hz, 1H, NHCbz), 5.15 (ddd, ${}^{3}J_{HH}$ 3.0, 8.3 Hz, ${}^{3}J_{\text{HP}}$ 11.5 Hz, 1H, CHNHCbz), 5.09–5.00 (m, 2H, CH₂Ph), 4.17–3.90 (m, 4H, 2CH₃CH₂O), 3.39 (dd, ³J_{HH} 3.0 Hz, ²J_{HP} 16.0 Hz, 1H, CHNH₂), 1.65 (br s, 2H, NH₂), 1.26, 1.20 (2t, 3 J_{HH} 7.1 Hz, 6H, 2CH₃CH₂O); δ_C (63 MHz, CDCl₃) 155.6 (s, C=O), 139.8 (d, ³J_{CP} 5.0 Hz, CHC_{Ar}), 136.4 (C_{Ar}), 128.3, 128.2, 127.9, 127.8, 127.3, 126.4, (CH_{Ar}), 66.4 (s, CH₂Ph), 62.4, 62.2 (2d, ²J_{CP} 9.4 Hz, 2CH₃CH₂O), 55.1 (s, CHNHCbz), 53.3 (d, 1 J_{CP} 152.3 Hz, CHNH₂), 16.2, 16.1 (2d, 3 J_{CP} 5.3 Hz, 2CH3CH2O).

4.3.8. Diethyl syn/anti-(1-amino-2-benzyloxycarbonylamino-propyl) phosphonates 4h. Crude product was purified by flash chromatography (AcOEt) to give the *title compound* $4h$ (60 mg, 58%) as a colorless oil; [found: C, 52.53; H, 7.35; N, 8.16. $C_{15}H_{25}N_2O_5P$ requires C, 52.32; H, 7.32; N, 8.14%]; R_f (AcOEt) 0.14; v_{max} (neat) 3296, 2979, 1710, 1530, 1224, 1018, 957, 696; δ_P (101 MHz, CDCl₃) 26.97 (syn-4h 59%), 26.78 (anti-4h 41%); δ_H (250 MHz, CDCl₃) 7.36-7.24 (m, 5HAr, syn and anti), 6.00-5.87 (m, 1H, NHCbz, syn and anti), 5.08 (s, 2H, CH₂Ph, syn and anti), 4.15-3.97 (m, 5H, 2CH₃CH₂O, CHNHCbz, syn and anti), 3.18 (dd, 3 J_{HH} 3.7 Hz, 2 J_{HP} 16.1 Hz, 1H, CHNH₂, anti), 3.05 (dd, 3 J_{HH} 3.7 Hz, 2 J_{HP} 15.2 Hz, 1H, CHNH₂, syn), 1.89 (br s, 2H, NH₂, syn and anti), 1.41-1.15 (m, 9H, 2CH₃CH₂O, CH₃CH, syn and anti); δ_c (63 MHz, CDCl₃) 154.0, 153.9 (2s, C=O, syn and anti), 134.9, 134.8 (2s, C_{Ar}, syn and anti), 126.6, 126.5, 126.1 (3s, CHAr, syn and anti), 64.6 (s, CH₂Ph, syn and anti), 60.6, 60.3 (2d, 2 J_{CP} 6.5, 6.1 Hz, CH₃CH₂O, syn and anti), 51.0 (d, 1 J_{CP} 148.5 Hz, CHNH₂, syn), 50.2 (d, ¹J_{CP} 144.5 Hz, CHNH₂, anti), 46.6 (s, CHNHCbz, syn), 46.0 (d, $^{2}J_{CP}$ 9.1 Hz, CHNHCbz, anti), 14.6 (d, $\mathrm{^{3}J_{CP}}$ 5.7 Hz, 2CH₃CH₂O, syn and *anti*), 14.2 (s, CH₃CH, syn and anti).

4.4. Diethyl trans-(5-phenyl-2-thioxoimidazolidin-4-yl) phosphonate^{[21k](#page-8-0)} 5a

Trifluoroacetic acid (0.9 mL, 12 mmol) was added dropwise to the solution of syn-4a, (100 mg, 0.27 mmol) in CH_2Cl_2 (1.5 mL). The mixture was stirred for 1.5 h at room temperature, and the volatile materials evaporated under reduced pressure. The oily residue was dissolved in $CH₂Cl₂$ (1.0 mL), triethylamine (0.19 mL, 1.35 mmol) was added to the solution and the mixture was stirred for 30 min at room temperature. Then, the solution of 1,1'thiocarbonyldiimidazole (56 mg, 1.35 mmol) in CH_2Cl_2 (1.5 mL) was added and the mixture was stirred for 24 h at room temperature. Reaction mixture was diluted with CH_2Cl_2 (30 mL), and washed successively with 1 M aq HCl $(2\times3$ mL), saturated NaHCO₃ (2×3 mL), and brine (3 mL). Organic phase was dried over MgSO4, evaporated, and subjected to flash chromatography on silica gel (eluent: $CH_2Cl_2/$ acetone 3:1 v/v) to give pure trans-**5a** (43 mg, 51%) as a colorless solid, mp 151-153 °C; [found: C, 49.41; H, 5.80; N, 8.69. C₁₃H₁₉N₂O₃PS requires C, 49.67; H, 6.09; N, 8.91%]; R_f (CH₂Cl₂-acetone 3:1) 0.36; ν_{max} (ATR) 3144, 2904, 1532, 1472, 1232, 1024, 780; δ_P (101 MHz, CDCl₃) 18.46; δ_H (250 MHz, CDCl₃) 7.42-7.30 (m, 5H, H_{Ar}), 7.03 (br s, 2H, 2 NH), 5.21 (dd, 3 J_{HH} 6.3 Hz, 3 J_{HP} 20.5 Hz, 1H, CHC_{Ar}), 4.32–4.05 (m, 4H, CH3CH2O), 4.00 (d, 3 J_{HH} 6.3 Hz, 1H, CHP), 1.35, 1.31 (2t, 3 J_{HH} 7.1 Hz, 6H, CH₃CH₂O); δ _C (63 MHz, CDCl₃) 183.2 (d, ³J_{CP} 4.8 Hz, C=S), 139.9 (d, ³ J_{CP} 10.1 Hz, CHC_{Ar}), 129.0, 128.8, 126.1 (s, CH_{Ar}), 63.8 (d, ² L_n 6.8 Hz, CH₂CH₂O), 61.8 (s $J_{\rm CP}$ 6.8 Hz, CH₃CH₂O), 63.3 (d, $^2J_{\rm CP}$ 7.2 Hz, CH₃CH₂O), 61.8 (s, CHC_{Ar}), 61.6 (d, 1 J_{CP} 161.7 Hz, CHP), 16.5 (d, 3 J_{CP} 5.3 Hz, CH₃CH₂O), 16.4 (d, 3 J_{CP} 5.7 Hz, CH₃CH₂O).

4.5. Diethyl trans/cis-(5-ethyl-2-thioxoimidazolidin-4-yl) phosphonates 21k 5b

A crude mixture of trans- and cis-5b was obtained in the same way as 5a. Crude product was purified by flash chromatography $(CH_2Cl_2/$ acetone 1:1) to give the title trans- and cis-**5b** (29 mg, 41%) as yellow solid, mp 120-123 °C; [found: C, 40.74; H, 7.38; N, 10.18. C₉H₁₉N₂O₃PS requires C, 40.59; H, 7.19; N, 10.52%]; Rf (CH₂Cl₂/acetone 1:1) 0.45; ν_{max} (ATR) 3163, 2967, 1523, 1225, 1011, 965, 719, 624; δ_P (101 MHz, CDCl₃) 19.24 (trans-5b 64%), 18.21 (cis-5b 36%); δ_H (250 MHz, CDCl₃) 7.27, 7.21 (2br s, 1H, NH, cis), 6.83, 6.57 (2br s, 1H, NH, trans), $4.31-4.04$ (m, 6H, 2CH₃CH₂O, CH₃CH₂CH, trans and cis, CHP, cis), 3.83 (br d, 3 J_{HH} 6.4 Hz, 1H, CHP, trans), $2.03-1.77$ (m, 2H, CH₃CH₂CH, cis), 1.75–1.61 (m, 2H, CH₃CH₂CH, trans), 1.36, 1.35 (2t, 6H^{*}, ³J_{HH} 7.1 Hz, 2CH₃CH₂O, trans and cis), 1.01 (t, 3H*, 3 J_{HH} 7.5 Hz, CH₃CH₂, cis), 0.97 (t, 3H*, 3 J_{HH} 7.4 Hz, CH₃CH₂, trans); δ_{C} (63 MHz, CDCl₃) 183.3 (d, $3J_{CP}$ 3.7 Hz, C=S, cis), 182.2 (d, $3J_{CP}$ 4.3 Hz, C=S, trans), 63.6 (d, $\frac{2}{\text{C}}$ 6.8 Hz, CH₃CH₂O, trans), 63.2 (d, $\frac{2}{\text{C}}$ 6.1 Hz, CH₃CH₂O, trans), 63.1 (d, ²J_{CP} 7.3, CH₃CH₂O, trans), 63.0 (d, ²J_{CP} 7.4, CH₃CH₂O, cis), 61.1 (s, CH₃CH₂CH, cis), 60.0 (s, CH₃CH₂CH trans), 57.8 (d, 1 J_{CP} 163.2 Hz, CHP, trans), 56.7 (d, 1 J_{CP} 157.3 Hz, CHP, cis), 28.7 (d, 3 J_{CP} 10.7 Hz, CH₃CH₂CH, trans), 23.9 (d, 3 J_{CP} 7.0 Hz, CH₃CH₂CH, cis), 16.4, 16.3 (2br d, ³J_{CP} 5.9 Hz, CH₃CH₂O, trans and cis), 11.4 (s, CH_3CH_2CH , cis), 9.0 (s, CH_3CH_2CH , trans).

4.6. Enantioselective synthesis of diethyl syn-(1-amino-2 tert-butoxycarbonylamino-2-phenylethyl)phosphonates 4a'

Solid, anhydrous K_2CO_3 (700 mg, 5.0 mmol) was added to the solution of diethyl nitromethanephosphonate (1, 197 mg, 1.0 mmol), α -amido sulfone **2a** (379 mg, 1.05 mmol), and chiral thiourea 7 (120 mg, 0.20 mmol) in anhydrous toluene (10 mL). The suspension was vigorously stirred for 3 d at room temperature. Then, the mixture was carefully quenched with 10% aq KHSO₄ (10 mL). The organic layer was separated and the aqueous phase was extracted with CH_2Cl_2 (3×15 mL). The combined organic phases were washed with brine (2×10 mL), dried over MgSO₄, evaporated, and subjected to flash chromatography on silica gel (AcOEt/hexanes 5:2 v/v) to give 310 mg (71%) of pure syn- and antiadducts $3a'$ (syn/anti=57/43). Pure enantioenriched syn-3a' (109 mg) was obtained in 27% yield after crystallization from CCl4 ([Table 3\)](#page-3-0). Compound syn- $3a'$ (100 mg, 0.247 mmol) was subjected to $NabH_4/NiCl_2$ system in methanol according to the general procedure described in Section [4.3](#page-5-0) to give, after reduction and flash chromatography, 56 mg (61%) of syn-**4a'**, [α] $^{20}_{D}$ –9.2 (c 0.15, CHCl₃,

for a sample with 47% ee). The spectroscopic data of syn- $3a'$ and syn-4a' were identical to racemic syn-3a and syn-4a.

Acknowledgements

Financial support by a grant 1T 09A 054 30 $(2006-2009)$ from the Ministry of Education and Science is gratefully acknowledged.

Supplementary data

Supplementary data associated with this article can be found in the online version at doi:10.1016/j.tet.2010.10.072.

References and notes

- 1. For recent reviews, see: (a) Shibasaki, M.; Kanai, M. Chem. Pharm. Bull. 2001, 49, 511; (b) Westermann, B. Angew. Chem., Int. Ed. 2003, 42, 151; (c) Vilaivan, T.; Bhanthumnavin, W.; Sritana-Anant, Y. Curr. Org. Chem. 2005, 9, 1315; (d) Petrini, M.; Torregani, E. Synthesis 2007, 159; (e) Ting, A.; Schaus, S. E. Eur. J. Org. Chem. 2007, 5797; (f) Marqués-López, E.; Merino, P.; Tejero, T.; Herrera, R. P. Eur. J. Org. Chem. 2009, 2401; (g) Arrayás, R. G.; Carretero, J. C. Chem. Soc. Rev. 2009, 38, 1940.
- 2. For recent examples, of the aza-Henry reaction using chiral betaine and guanidine catalysts, see: (a) Uraguchi, D.; Koshimoto, K.; Ooi, T. J. Am. Chem. Soc. 2008, 130, 10878; (b) Lovick, H. M.; Michael, F. E. Tetrahedron Lett. 2009, 50, 1016.
- 3. For recent examples, of the aza-Henry reaction using chiral Lewis acids and chiral proton catalysts, see: (a) Nishiwaki, N.; Gothelf, A. S.; Knudsen, R.; Jørgensen, K. A. Angew. Chem., Int. Ed. 2001, 40, 2992; (b) Nishiwaki, N.; Gothelf, A. S.; Knudsen, R.; Risgaard, T.; Jørgensen, K. J. Am. Chem. Soc. 2001, 123, 5843; (c) Knudsen, K. R.; Jørgensen, K. A. Org. Biomol. Chem. 2005, 3, 1362; (d) Lee, A. M.; Kim, W.; Lee, J.; Hyeon, T.; Kim, B. M. Tetrahedron: Asymmetry 2004, 15, 2595; (e) Palomo, C.; Oiarbide, M.; Halder, R.; Laso, A.; López, R. Angew. Chem., Int. Ed. 2006, 45, 117; (f) Trost, B. M.; Lupton, D. W. Org. Lett. 2007, 9, 2023; (g) Gao, F.; Zhu, J.; Tang, Y.; Deng, M.; Qian, C. Chirality 2006, 18, 741; (h) Anderson, J. C.; Blake, A. J.; Howell, G. P.; Wilson, C. J. Org. Chem. 2005, 70, 549; (i) Anderson, J. C.; Howell, G. P.; Lawrence, R. M.; Wilson, C. S. J. Org. Chem. 2005, 70, 5665; (j) Zhou, H.; Peng, D.; Qin, B.; Hou, Z.; Liu, X.; Feng, X. J. Org. Chem. 2007, 72, 10302; (k) Nugent, B. M.; Yoder, R. A.; Johnston, J. N. J. Am. Chem. Soc. 2004, 126, 3418; (l) Singh, A.; Yoder, R. A.; Shen, B.; Johnston, J. N. J. Am. Chem. Soc. 2007, 129, 3466; (m) Shen, B.; Johnston, J. N. Org. Lett. 2008, 10, 4397; (n) Singh, A.; Johnston, J. N. J. Am. Chem. Soc. **2008**, 130, 5866; (o) Rueping, M.; Antonchick, A. P. Org. Lett. 2008, 10, 1731; (p) Davis, T. A.; Wilt, J. C.; Johnston, J. N. J. Am. Chem. Soc. 2010, 132, 2880; (q) Hess, A. S.; Yoder, R. A.; Johnston, J. N. Synlett 2006, 147.
- 4. For recent examples, of the aza-Henry reaction using chiral thioureas catalysts, see: (a) Yoon, T. P.; Jacobsen, E. N. Angew. Chem., Int. Ed. 2005, 44, 466; (b) Xu, X.; Furukawa, T.; Okino, T.; Miyabe, H.; Takemoto, Y. Chem.-Eur. J. 2006, 12, 466; (c) Han, B.; Liu, Q.-P.; Li, R.; Tian, X.; Xiong, X.-F.; Deng, J.-G.; Chen, Y.-C. Chem.—Eur. J. 2008, 14, 8094; (d) Wang, C.; Zhou, Z.; Tang, C. Org. Lett. 2008, 10, 1707; (e) Wang, C.; Dong, X.; Zhang, Z.; Xue, Z.; Teng, H. J. Am. Chem. Soc. 2008, 130, 8606; (f) Takada, K.; Nagasawa, K. Adv. Synth. Catal. 2009, 351, 345; (g) Okino, T.; Nakamura, S.; Furukawa, T.; Takemoto, Y. Org. Lett. 2004, 6, 625; (h) Robak, M. T.; Trincado, M.; Ellman, J. A. J. Am. Chem. Soc. 2007, 129, 15110; (i) Jiang, X.; Zhang, Y.; Wu, L.; Zhang, G.; Liu, X.; Zhang, H.; Fu, D.; Wang, R. Adv. Synth. Catal. 2009, 351, 2096; (j) Rampalakos, C.; Wulff, W. D. Adv. Synth. Catal. 2008, 350, 1785; (k) Chang, Y.; Yang, J.; Dang, Y.; Xue, Y. Synlett 2007, 2283.
- 5. For recent examples, of the aza-Henry reaction using cinchona alkaloids catalysts, see: (a) Palomo, C.; Oiarbide, M.; Laso, A.; López, R. J. Am. Chem. Soc. 2005 127, 17622; (b) Fini, F.; Sgarzani, V.; Pettersen, D.; Herrera, R. P.; Bernardi, L.; Ricci, A. Angew. Chem., Int. Ed. 2005, 44, 7975; (c) Bernardi, L.; Fini, F.; Herrera, R. P.; Ricci, A.; Sgarzani, V. Tetrahedron 2006, 62, 375; (d) Gomez-Bengoa, E.; Linden, A.; López, R.; Múgica-Mendiola, I.; Oiarbide, M.; Palomo, C. J. Am. Chem. Soc. 2008, 130, 7955; (e) Bode, C. M.; Ting, A.; Schaus, S. E. Tetrahedron 2006, 62, 11499.
- 6. For recent examples of the aza-Henry reaction using heterobimetallic catalysts, see: (a) Yamada, K.; Moll, G.; Shibasaki, M. Synlett 2001, 980; (b) Handa, S.; Gnanadesikan, V.; Matsunaga, S.; Shibasaki, M. J. Am. Chem. Soc. 2010, 132, 4925; (c) Yamada, K.; Harwood, S. J.; Gröger, H.; Shibasaki, M. Angew. Chem., Int. Ed. 1999, 38, 3504; (d) Handa, S.; Gnanadesikan, V.; Matsunaga, S.; Shibasaki, M. J. Am. Chem. Soc. 2007, 129, 4900; (e) Chen, Z.; Morimoto, H.; Matsunaga, S.; Shibasaki, M. J. Am. Chem. Soc. 2008, 130, 2170.
- 7. For application of N-formyl and N-Carbamate protected α -aminoalkyl-arylsulfones in the aza-Henry reaction, see: (a) Ballini, R.; Petrini, M. Tetrahedron Lett. 1999, 40, 4449; (b) Petrini, M.; Torregiani, E. Tetrahedron Lett. 2006, 47, 3501; (c) Foresti, E.; Palmieri, G.; Petrini, M.; Profeta, M. Org. Biomol. Chem. 2003, 1, 4275; (d) Nagano, T.; Kinoshita, H. Bull. Soc. Chem. Jpn. 2000, 73, 1605; (e) Shiraishi, Y.; Yamauchi, H.; Takamura, T.; Kinoshita, H. Bull. Soc. Chem. Jpn. 2004, 77, 2219; (f) Refs. 4i,5a,b,d.
- 8. Adams, H.; Anderson, J. C.; Peace, S.; Pennell, A. M. K. J. Org. Chem. 1998, 63, 9932.
- 9. For a recent review on the synthesis of α , β -diamino acids, see: Viso, A.; de la Pradilla, R. F.; Garcia, A.; Flores, A. Chem. Rev. 2005, 105, 3167.
-
- 10. Wilt, J. C.; Pink, M.; Johnston, J. N. *Chem. Commun.* **2008**, 4177.
11. Baranov, G. M.; Perekalin, V. V. Russ. *Chem. Rev.* **1992**, 61, 1220.
- 12. Petrov, K. A.; Chauzov, V. A.; Bogdanov, N. N.; Agafonov, S. V. Zh. Obshch. Khim. 1979, 49, 90.
- 13. Petrov, K. A.; Chauzov, V. A.; Bogdanov, N. N.; Pastukhova, I. V. J. Gen. Chem. USSR 1976, 46, 1230.
- 14. For recent review, see: Petrini, M. Chem. Rev. 2005, 105, 3949.
- 15. For the synthesis of N-Boc and N-Cbz protected a-amido sulfones, see: (a) Engeberts, J. B. F. N.; Strating, J. *Recl. Trav. Chim. Pays-Bas* **1964**, 83, 733; (b)
Pearson, W. H.; Lindbeck, A. C.; Kampf, J. W. *J. Am. Chem. Soc.* **1993**, 115, 2623; (c) Kanazawa, A. M.; Denis, J.-N.; Greene, A. E. J. Org. Chem. 1994, 59, 1238; (d) Mecozzi, T.; Petrini, M. J. Org. Chem. 1999, 64, 8970; (e) Murry, J. A.; Frantz, D. E.; Soheili, A.; Tillyer, R.; Grabowski, E. J.; Reider, P. J. Am. Chem. Soc. 2001, 123, 9696; (f) Bernacka, E.; Klepacz, A.; Zwierzak, A. Tetrahedron Lett. 2001, 42, 5093; σ) Klepacz, A.; Zwierzak, A. Tetrahedron Lett. 2002, 43, 1079.
- 16. For direct conversion of N-carbamate protected a-amido sulfones into N-carbamate imines, see: (a) Song, J.; Wang, Y.; Deng, L. J. Am. Chem. Soc. 2006, 128, 6048; (b) Trost, B. M.; Jaratjaroonphong, J.; Reutrakul, V. J. Am. Chem. Soc. 2006, 128, 2778; (c) Wenzel, A. G.; Jacobsen, E. N. J. Am. Chem. Soc. 2002, 124, 12964.
- 17. For recent reviews on asymmetric PTC, see: (a) Albanese, D. Mini-Rev. Org. Chem. 2006, 3, 195; (b) Hashimoto, T.; Maruoka, K. Chem. Rev. 2007,107, 5656; (c) Ooi, T.; Maruoka, K. Angew. Chem., Int. Ed. 2007, 46, 4222; (d) Maruoka, K. Org. Process Res. Dev. 2008, 12, 679; (e) Jew, S.; Park, H. Chem. Soc. Rev. 2009, 7090.
- 18. Zon, J. Synthesis 1984, 661.
- 19. Lejczak, B.; Kafarski, P.; Zygmunt, J. Biochemistry 1989, 28, 3549.
- 20. Palacios, F.; Alonso, C.; De Los Santos, J. Asymmetric Synthesis of a-Substitutedb-Amino Phosphonates and Phosphinates and b-Amino Sulfur Analogs In Enantioselective synthesis of β -amino acids; Juaristi, E., Soloshonok, V. A., Eds.; John Wiley: Hoboken, NJ, 2005; pp 292-294.
- 21. For recent syntheses of 1,2-diaminophosphonates, see: (a) Zygmunt, J. Tetrahedron 1985, 41, 4979; (b) Dolence, E. K.; Roylance, J. B. Tetrahedron: Asymmetry 2004, 15, 3307; (c) Li, B.-F.; Zhang, M.-J.; Hou, X.-L.; Dai, L.-X. J. Org. Chem. 2002, 67, 2902; (d) Piotrowska, D. G.; Wróblewski, A. Tetrahedron 2003, 59, 8405; (e) Dolence, E. K.; Mayer, G.; Kelly, B. D. Tetrahedron: Asymmetry 2005, 16, 1583; (f) De Risi, C.; Dondoni, A.; Perrone, D.; Pollini, G. P. Tetrahedron Lett. 2001, 42, 3033; (g) De Risi, C.; Perrone, D.; Dondoni, A.; Pollini, G. P.; Bertolasi, V. Eur. J. Org. Chem. 2003, 1904; (h) Studer, A.; Seebach, D. Heterocycles 1995, 40, 357; (i) Campbell, M. M.; Carruthers, N. J. Chem. Soc., Chem. Commun. 1980, 730; (j) Campbell, M. M.; Carruthers, N. I.; Mickel, S. J. Tetrahedron **1982**, 38, 2513; (k) Btaszczyk, R.; Gaida, T. Tetrahedron Lett. 2007, 48, 5859; (1) Louaisil, N.; Rabasso, N.; Fadel, A. Synthesis 2007, 289; (m) Kobayashi, S.; Yazaki, R.; Seki, K.; Yamashita, Y. Angew. Chem., Int. Ed. 2008, 47, 5613; (n) Momo, R. D.; Fini, F.; Bernardi, L.; Ricci, A. Adv. Synth. Catal. 2009, 351, 2283.
- 22. Nose, A.; Kudo, T. Chem. Pharm. Bull. 1981, 29, 1159.
- 23. (a) Sohtome, Y.; Hashimoto, Y.; Nagasawa, K. Eur. J. Org. Chem. 2006, 2894; (b) Takada, K.; Takemura, N.; Cho, K.; Sohtome, Y.; Nagasawa, K. Tetrahedron Lett. 2008, 48, 1623.
- 24. Nudelman, A.; Marcovici-Mizrahi, D.; Nudelman, A.; Flint, D.; Wittenbach, V. Tetrahedron 2004, 60, 1731.
- 25. Makowski, M.; Rzeszotarska, B.; Smelka, L.; Kubica, Z. Liebigs Ann. Chem. 1985, 775.
- 26. Davies, S. G.; Evans, G. B.; Pearce, S. Tetrahedron 1994, 50, 7521.
- 27. Błażewska, K.; Gajda, T. Tetrahedron: Asymmetry 2002, 13, 671.
- 28. Kazuo, N.; Shinji, Y.; Takao, A.; Keiichi, I. Synth. Commun. 1990, 20, 2033.
- 29. (a) Błażewska, K.; Paneth, P.; Gajda, T. J. Org. Chem. 2007, 72, 878; (b) Błażewska, K. D.; Gajda, T. *Tetrahedron: Asymmetry 2009, 20, 1337.*
30. Vakulya, B.; Varga, S.; Csámpai, A.; Soós, T. *Org. Lett. 2005*, 7, 1967.
-
- 31. For a recent paper on the absolute configuration determination of vicinal-diamines using NMR spectroscopy and O-methyl mandelic acid as CDA, see: [Ref. 6b.](#page-7-0)